## Amendments to the Claims:

The following listing of claims will replace all prior versions, and listings, of claims in the application:

- 1. (Original) A method of producing cerium oxide particles by raising a temperature of a cerium compound from a normal temperature and heating the cerium compound to a temperature range of 400°C to 1200°C, in which the method comprises at least a temperature raising stage of a temperature rise speed of 2°C/hour to 60°C/hour.
- 2. (Original) The production method for cerium oxide particles according to claim 1, wherein the temperature raising stage of the temperature rise speed of 2°C/hour to 60°C/hour is a first temperature raising stage that is continued until the temperature reaches a temperature range of 200°C to 350°C after rising from the normal temperature.
- 3. (Original) The production method for cerium oxide particles according to claim 2, wherein after the first temperature raising stage, heating is performed up to the temperature range of 400°C to 1200°C in a second temperature raising stage of a temperature rise speed of 2°C/hour to 200°C/hour.
- 4. (Currently Amended) The production method for cerium oxide particles according to any one of claims 1 to 3, wherein after the temperature reaches the temperature range of 400°C to 1200°C, the temperature is maintained for 10 minutes to 240 hours.

- 5. (Currently Amended) A cerium oxide powder made of cerium oxide particles produced in a ceramics-made container based on the method according to any one of claims 1 to 4, in which a difference in BET method-converted particle diameter converted from a specific surface area value obtained through a BET method between cerium oxide particles of a surface layer portion and an internal portion of the powder within the container is within 20% of an average value of BET method-converted particle diameters of cerium oxide particles of the entire powder within the container.
- 6. (Currently Amended) A cerium oxide powder made of cerium oxide particles produced in plural ceramics-made containers based on the method according to any one of claims 1 to 4, in which a standard deviation of BET method-converted particle diameters converted from specific surface area values obtained through a BET method, and an average value of the BET method-converted particle diameters are such that the value calculated through the following expression I:

[(standard deviation)/(average value)] × 100 (I)
in the expression, (standard deviation) represents a standard deviation of BET methodconverted particle diameters, and (average value) represents an average value of BET
method-converted particle diameters,
is within a range of 3 and 10.

7. (Original) A method of producing cerium oxide particles by raising a temperature of a cerium compound from a normal temperature and heating the cerium compound to a temperature range of 400°C to 1200°C, in which the method proceeds via a stage of heating while supplying a humidified gas in a temperature raising process.

8. (Original) The production method for cerium oxide particles according to claim 7, wherein a water vapor in the humidified gas has a value of 0.5 to 0.8 in a partial pressure ratio calculated through the following expression II:

$$H_2O_p/(H_2O_p + gas_p)$$
 (II)

in the expression, H<sub>2</sub>O<sub>p</sub> represents the partial pressure of water vapor, and gas<sub>p</sub> represents the partial pressure of the gas.

- 9. (Currently Amended) The production method for cerium oxide particles according to claim 7 or 8, wherein the gas is an oxygen gas, a mixture gas of oxygen and nitrogen, or an air.
- 10. (Currently Amended) The production method for cerium oxide particles according to any one of claims 7 to 9, wherein supply of the humidified gas is started at a temperature of 100°C or higher, and is continued until a temperature range of 200°C to 350°C is reached.
- 11. (Currently Amended) The production method for cerium oxide particles according to any one of claims 7 to 10, wherein the cerium compound is a cerium carbonate hydrate.
- 12. (Currently Amended) A cerium oxide powder made of cerium oxide particles produced in an atmosphere adjustment type calcining furnace based on the method according to any one of claims 7 to 11, in which values of particle diameters of the powder measured through a laser diffraction method become 0.1 to 0.3 as a ratio value calculated through the following expression III:

$$(D_{50} - D_{10})/(D_{90} - D_{50})$$
 (III)

in the expression,

 $D_{10}$  represents a particle diameter that means that the number of particles of this particle diameter or less is 10% of the total number of particles,

 $D_{50}$  represents a particle diameter that means that the number of particles of this particle diameter or less is 50% of the total number of particles, and

 $D_{90}$  represents a particle diameter that means that the number of particles of this particle diameter or less is 90% of the total number of particles.

- 13. (Currently Amended) An aqueous cerium oxide slurry for use for the polishing of a substrate whose main component is silica, which contains cerium oxide particles produced by the method according to any one of claims 1 to 4 and claims 7 to 11.
- 14. (Original) The aqueous cerium oxide slurry according to claim 13, wherein the polishing of the substrate whose main component is silica is the polishing of a rock crystal, a photomask-purpose quartz glass, an organic film for a semiconductor device, a low-dielectric constant film for a semiconductor device, an inter-layer insulator film for a semiconductor device, trench isolation for a semiconductor device, or a glass-made hard disk substrate.